

Protein Side Chain Absorptions in the Infrared Finger Print Region

Excerpted from: Lauren DeFlores, *Multi-mode Vibrational Spectroscopy of Peptides and Proteins*, Ch. 6 (PhD Thesis, Massachusetts Institute of Technology, 2008)

See also:

Andreas Barth and Christian Zscherp, "What vibrations tell us about proteins," *Quarterly Reviews of Biophysics* **35** (2002) 369–430

Andreas Barth, "The infrared absorption of amino acid side chains," *Progress in Biophysics & Molecular Biology* **74** (2000) 141–173

Table 1. Extinction coefficients and peak frequencies of amino acids that absorb between 1200 cm⁻¹ and 1800 cm⁻¹ in H₂O and D₂O. Frequencies are tabulated as a function of pH relative to the pK_a value determined from the isolated amino acid. For vibrations with no pH dependence appear in the central column.

Table 1a. Side Chain Absorptions in H₂O

Amino Acid	cm ⁻¹ (H ₂ O) < pK _s low pH		cm ⁻¹ (H ₂ O)		cm ⁻¹ (H ₂ O) > pK _s high pH		Mode	pK _s
Arginine ARG R	460	1652	-	-	-	-	v _{as} CN ₃ H ₅ ⁺	11.6 - 12.6
	320	1630	-	-	-	-	v _s CN ₃ H ₅ ⁺	
Aspartic Acid ASP D	-	-	280	1716	-	-	vC=O	4.0 - 4.8
	-	-	-	-	235	1577	v _{as} COO ⁻	
	-	-	-	-	256	1402	v _s COO ⁻	
	-	-	-	1375	-	-	δ _s CH ₃	
Asparagine ASN N	-	-	320	1677	-	-	vC=O	
	-	-	150	1617	-	-	δNH ₂	
Cysteine CYS C	-	-	-	2551	-	-	vSH	9.0-9.5
Glutamic Acid GLU E	-	-	220	1712	-	-	vC=O	4.4 - 4.6
	-	-	-	-	460	1558	v _{as} COO ⁻	
	-	-	-	-	316	1404	v _s COO ⁻	
Glutamine GLN Q	-	-	370	1680	-	-	vC=O	
	-	-	230	1595	-	-	δNH ₂	
	-	-	-	1410	-	-	vCN	
Histidine HIS H	250	1631	-	-	-	-	vC=C (H ₂ ⁺)	6.0-7.0
	70	1575, 1594	-	-	-	-	vC=C (H)	
	-	-	-	-	-	1439	δCH ₃ , vCN (')	
Lysine LYS K	80	1626	-	-	-	-	δ _{as} NH ₃ ⁺	10.4-11.1
	85	1526	-	-	-	-	δ _s NH ₃ ⁺	
Phenylalanine PHE F	-	-	80	1494	-	-	vCC ring	
	-	-	-	1460	-	-	δ _{as} CH ₃	
Proline PRO P	-	-	-	1432	-	-	vCN	
	-	-	-	1450	-	-	δCH ₂	
Tryptophan TRP W	-	-	-	1622	-	-	vCC, vC=C	
	-	-	-	1509	-	-	vCN, δCH, δNH	
	-	-	-	1496	-	-	vCC, δCH	
	-	-	-	1462	-	-	δCH, vCC, vCN	
	-	-	-	1427	-	-	dNH, vCC, δCH	
Tyrosine TYR Y	120	1617	-	-	-	-	vCC, vCH	9.8 - 10.4
	85	1598	-	-	160	1601	vCC	
	385	1515	-	-	-	-	vCC, δCH	
	-	-	-	-	700	1499	vCC, δCH	
	-	-	-	-	580	1270	vCO, δCC	
	200	1250	-	-	-	-	vCO, δCC	

Table 1b. Side Chain Absorptions in D₂O

Amino Acid	cm ⁻¹ (D ₂ O) < pK _s low pH		cm ⁻¹ (D ₂ O)		cm ⁻¹ (D ₂ O) > pK _s high pH		Mode	pK _s (pH)
Arginine ARG R	460	1605	-	-	-	-	v _{as} CN ₃ D ₅ ⁺	11.6 - 12.6
	500	1586	-	-	-	-	v _s CN ₃ D ₅ ⁺	
Aspartic Acid ASP D	-	-	290	1713	-	-	vC=O	4.0 - 4.8
	-	-	-	-	820	1584	v _{as} COO ⁻	
	-	-	-	-	-	1404	v _s COO ⁻	
Asparagine ASN N	-	-	570	1648	-	-	vC=O	
Cysteine CYS C	-	-	-	1849	-	-	vSD	9.0-9.5
Glutamic Acid GLU E	-	-	280	1706	-	-	vC=O	4.4 - 4.6
	-	-	-	-	830	1567	v _{as} COO ⁻	
	-	-	-	-	-	1407	v _s COO ⁻	
Glutamine GLN Q	-	-	550	1640	-	-	vC=O	
	-	-	-	1163	-	-	δND ₂	
	-	-	-	1409	-	-	vCN	
Histidine HIS H	35	1600	-	-	-	-	vC=C (D ₂ ⁺)	6.0-7.0
	70	1569, 1575	-	-	-	-	vC=C (D)	
	-	-	-	-	-	1439	δCD ₃ , vCN (')	
Lysine LYS K	-	1200	-	-	-	-	δ _{as} ND ₃ ⁺	10.4-11.1
	-	1170	-	-	-	-	δ _s ND ₃ ⁺	
Tryptophan TRP W	-	-	-	1618	-	-	vCC, vC=C	
	-	-	200	1455	-	-	δCD, vCC, vCN	
	-	-	-	1382	-	-	δND, vCC, δCD	
Tyrosine TYR Y	160	1615	-	-	-	-	vCC, vCD	9.8 - 10.4
	50	1590	-	-	350	1630	vCC	
	500	1515	-	-	-	-	vCC, δCD	
	-	-	-	-	650	1499	vCC, δCD	
	150	1255	-	-	-	-	vCO, δCC	

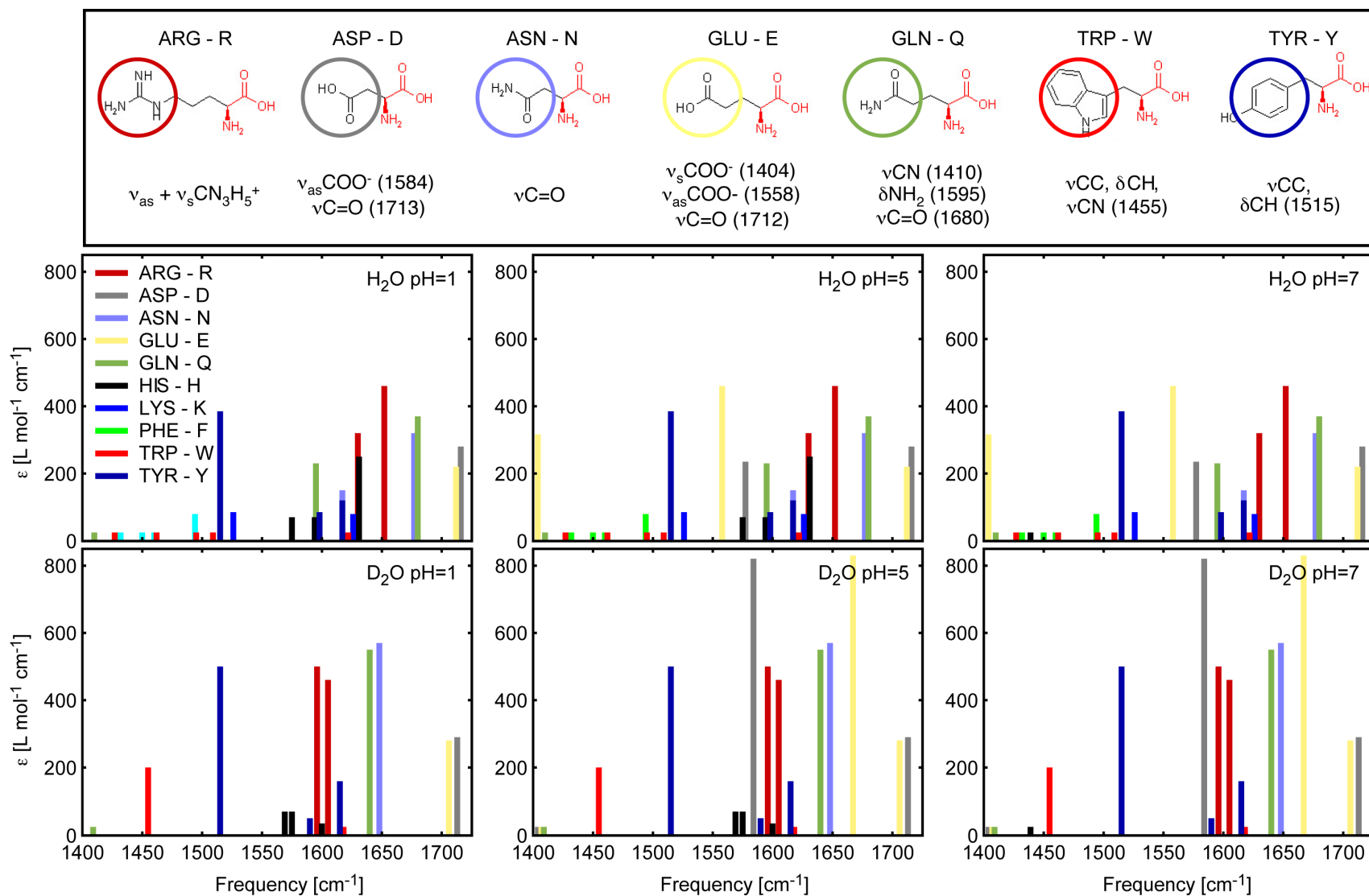


Figure 1. (Top) Atomic structures of predominant side chains in the amide finger region. (Bottom) Stick plots of side chain absorption from Table 1 in the amide finger print region as a function of solvent and pH. Major changes occur due to the protonation state of ASP, GLU and HIS. Isotopic sensitivity of the vibrational absorption is seen in TRP, ARG, GLU and ASP.